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**Tetrahedron Letters** 

journal homepage: www.elsevier.com/locate/tetlet



# Stereoselective total synthesis of polyrhacitide A

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#### ARTICLE INFO

### ABSTRACT

Article history: Received 13 January 2010 Revised 9 February 2010 Accepted 11 February 2010 Available online 16 February 2010

Keywords: Polyrhacitide A Ring-closing metathesis L-Malic acid Stereoselective total synthesis of polyketide lactone isolated from Chinese medicinal ant *Polyrhacis lam-ellidens* is described. Ring-closing metathesis followed by stereoselective intramolecular oxa-Michael addition reactions were used to construct the bicyclic lactone moiety in the molecule and L-malic acid was used as a chiral pool material © 2010 Elsevier Ltd. All rights reserved.

Ants from different parts of the world produce a wide range of interesting natural products having important biological functions and of considerable pharmacological interest.<sup>1</sup> Very recently Jiang and co-workers have isolated two new bicyclic lactones polyrhacitides A (1) and B (2)<sup>2</sup> (Fig. 1) from the Chinese medicinal ants Polyrhacis lamellidens, which have been used widely in China as traditional folk medicine for the treatment of rheumatoid arthritis and hepatitis.<sup>3</sup> The structure and absolute configuration of the molecules were determined by detailed spectroscopic studies. Although significant biological activities of polyrhacitides A (1) and B (2) remain unknown, extract of p. lamellidens showed significant analgesic and anti-inflammatory activities.<sup>4</sup> Therefore one might expect similar bioactivity of polyrhacitides A (1) and B (2). Due to their potential biological activities and unique structural features, as well as limited availability from the natural source (56.2 mg of A and 19.2 mg of B from 2 kg of ants) prompted us to develop an efficient strategy for their syntheses. While we were preparing this Letter, the first total synthesis of polyrhacitides A and B has appeared in the literature, where the catalytic asymmetric Overman esterification was utilized to install all the stereogenic centers.<sup>5</sup> In this Letter, we report the second stereoselective total synthesis of polyrhacitide A starting from L-malic acid.

Retrosynthetic analysis of polyrhacitide A revealed that the bicyclic lactone in the molecule could be constructed from compound **3** via an intramolecular oxa-Michael addition. The Michael acceptor that is, the  $\alpha$ , $\beta$ -unsaturated six-membered  $\delta$ -lactone in **3** could be synthesized from the bis-olefinic compound **4**, which in turn could be obtained from the known compound **5** derived from L-malic acid (Scheme 1).

Our synthesis (Scheme 2) started from compound **5**, which was prepared from **6** according to the reported procedure.<sup>6a,b</sup> Protection of the secondary hydroxyl group of **5** as its benzyl ether followed by acetonide deprotection under acidic conditions furnished diol **8**<sup>6c</sup> in 90% yield over two steps. Chemoselective tosylation of the primary alcohol followed by tosyl displacement with the vicinal hydroxyl group under basic conditions furnished epoxide **9**<sup>6c</sup> in 70%. Crucial opening of the epoxide **9** at the less substituted carbon atom with the anion generated from dithiane **9a** using *n*-butyllithium in THF under argon atmosphere gave compound **10** in 80% yield.<sup>7</sup> Deprotection of the dithioacetal group with AgNO<sub>3</sub> in aqueous ethanol at 60 °C furnished keto compound **11** in 78% yield.<sup>8</sup>

Hydroxyl directed stereoselective reduction of the keto group using NaBH<sub>4</sub> and Et<sub>2</sub>BOMe<sup>9</sup> in THF/MeOH (4:1) at -78 °C gave 1,3-syn diol 12 in 83% (dr 95:5). Acetonide protection of the 1,3diol using 2,2-dimethoxy propane in the presence of catalytic amount CSA yielded compound 13 in 95% yield. We planned to introduce the C4-OH center via asymmetric allylation, but when the aldehyde obtained from the olefinic compound 13 was subjected to asymmetric allylation,<sup>10</sup> using Brown's<sup>10a</sup> as well as Keck<sup>10b</sup> protocols gave poor yields ( $\sim 5\%$ ). This unexpected result forced us to do the substrate-controlled allylation.<sup>11</sup> Thus the aldehyde on treatment with allylmagnesium bromide in ether gave compound 14 (2.5:1) as inseparable diastereomeric mixtures. On acylation with acryloyl chloride compound **14** gave the separable diastereomers 4 (major) and 4a (minor). To establish the stereochemistry at C4 compound 4a was converted to compound 15a in three steps. In the <sup>13</sup>C NMR spectrum of compound **15a**, the methyls of the C4:C6-O-isopropylidene group resonating at  $\delta$ 24.6 and 24.8 and the ketal carbon at  $\delta$  100.1 confirmed their anti relationship according to the reports of Rychnovsky<sup>12a</sup> and



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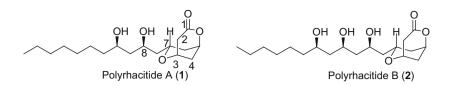
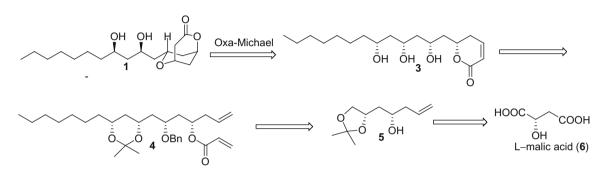


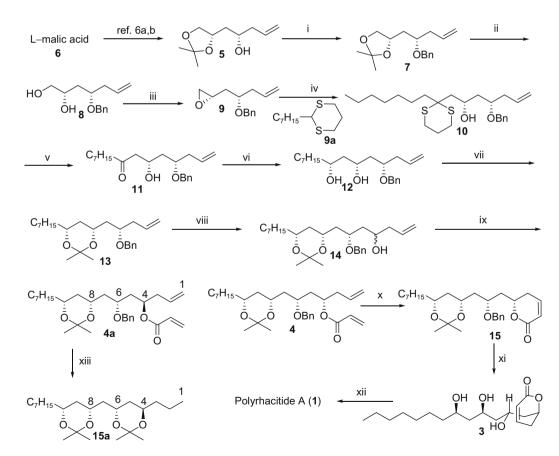
Figure 1. Structure of polyrhacitides A and B.

Evans.<sup>12b</sup> Therefore in the major isomer **4**, the C4-OH and C6-OH had 1,3-*syn* relationship. Next, the bis-olefinic compound **4** was subjected to RCM reaction using Grubbs 2nd generation catalyst<sup>13</sup>

in CH<sub>2</sub>Cl<sub>2</sub> to give compound **15**<sup>15a</sup> in 80% yield. Global deprotection of the compound **15** with TiCl<sub>4</sub> in CH<sub>2</sub>Cl<sub>2</sub> at 0 °C gave compound **3**, which on treatment with DBU<sup>14</sup> in THF at 0 °C afforded polyrhaci-



Scheme 1. Retrosynthetic analysis of polyrhacitide A.



**Scheme 2.** Reagents and conditions. (i) NaH, BnBr, TBAI (cat), DMF, 0 °C to rt, 24 h, 95%; (ii) AcOH, EtOH/H<sub>2</sub>O, (2.5:1.), 0 °C to rt, 24 h, 95%; (iii) (a) TsCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, DMAP, 0 °C to rt, 1 h; (b) K<sub>2</sub>CO<sub>3</sub>, MeOH, 0 °C to rt, 1 h, 70% over two steps; (iv) **9a**, nBuLi, THF, rt, 80%; (v) AgNO<sub>3</sub>, EtOH/H<sub>2</sub>O (10:1), 60 °C, 45 min, 78%; (vi) Et<sub>2</sub>BOMe, NaBH<sub>4</sub>, THF/ MeOH (4:1), -78 °C, 3 h, 83%, dr 95:5; (vii) 2,2-DMP, CSA (cat), CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to rt, 4 h, 95%; (viii) OsO<sub>4</sub>, NMO, acetone/H<sub>2</sub>O (1:1), 12 h; (b) NaIO<sub>4</sub>, THF/H<sub>2</sub>O (1:1), 0 °C to rt, 1 h; (c) allylmagnesium bromide, Et<sub>2</sub>O, 0 °C, 1 h; 60% over three steps, dr = 2.5:1; (ix) acryloyl chloride, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to rt, 30 min, 68% **4** and 27% **4a**; (x) Grubbs 2nd generation catalyst, CH<sub>2</sub>Cl<sub>2</sub>, 50 °C, 12 h, 80%; (xi) TiCl<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to rt, 30 min, 75%; (xii) DBU, THF, 0 °C to rt, 36 h, 90%; (xiii) (a) K<sub>2</sub>CO<sub>3</sub>, MeOH, rt, 1 h; (b) H<sub>2</sub>, Pd/C, MeOH; (c) 2,2-DMP, CSA (cat), CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to rt, 4 h, 75%.

tide A (1) in 72% over two steps, whose analytical data, (<sup>1</sup>H and <sup>13</sup>C)<sup>15b</sup> and the specific rotation (synthetic  $[\alpha]_D^{25} = +5.5$ , *c* 0.09, MeOH, reported  $[\alpha]_D^{25} = +8.3$ , *c* 0.6, MeOH) were in good agreement with the literature values.

In conclusion, we have achieved the stereoselective total synthesis of polyrhacitide A (1) from L-malic acid. Currently we are working on the total synthesis of polyrhacitide B, with a little modification of the above strategy, which will be reported in due course.

## Acknowledgments

We are thankful to DST, New Delhi, India for financial support (S.G.) and CSIR, New Delhi, India, (C.N.R.) for research fellowships. We are also thankful to Dr. J. S. Yadav Director of IICT, Dr. A. C. Kunwar, Dr. T. K. Chakraborty Director of CDRI for their support and encouragement and Professor Zhi-Hong Jiang for providing the copies of NMR (<sup>1</sup>H and <sup>13</sup>C) of natural polyrhacitide A.

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- 15. (a) Analytical data of compound **15**:  $R_{\rm f} = 0.5$  (silica gel, 20% EtOAc in hexane); [a)<sub>D</sub><sup>28</sup> -46.31 (c 0.19, CHCl<sub>3</sub>); IR (neat):  $\nu_{\rm max}$  2924, 2855, 1728, 1460, 1383, 1249, 1109, 741, 701 cm<sup>-1</sup>; <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  7.39–7.27 (m, 5H), 6.79 (ddd, J = 9.0, 6.5, 2.4 Hz, 1H), 6.0 (dd, J = 9.5, 2.4 Hz, 1H), 4.58 (m, 1H), 4.53 & 4.45 (two d, J = 11.8 Hz, 2H), 3.98 (m, 1H), 3.77 (pent, J = 5.9 Hz, 1H), 3.73 (m, 1H), 2.35 (t, J = 7.5 Hz, 2H), 2.14 (m, 1H), 1.88 (m, 1H), 1.63 (m, 1H), 1.4 (s, 3H), 1.36 (s, 3H), 1.26–1.24 (m, 15H), 0.88 (t, J = 7.0 Hz, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  164.5, 145.1, 138.3, 128.4, 128.1, 127.7, 121.3, 98.3, 75.3, 71.3, 70.3, 69.0, 65.6, 39.5, 38.8, 37.1, 36.4, 33.7, 32.0, 30.1, 29.7, 29.3, 29.2, 29.0, 24.9, 22.6, 14.1; MS (ESI): m/z (%) 459 (50) [M+H]<sup>+</sup>, 481 (100) [M+Na]<sup>+</sup>; HRMS (ESI): calcd for C<sub>28</sub>H<sub>42</sub>O<sub>5</sub> Na [M+Na]<sup>+</sup> 481.2909, found 481.2929.

(b) Analytical data for polyrhacitide A: Mp: 65 °C.  $R_f = 0.5$  (silica gel, 80% EtOAc in hexane); [z]<sub>D</sub><sup>28</sup> +5.5 (c 0.09, MeOH); IR (neat):  $\nu_{max}$  3421, 3315, 2923, 2854, 1729, 1454, 1340, 1201, 1087 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  4.88 (m, 1H), 4.40 (br s, 1H), 4.11–4.01 (m, 2H), 3.83 (m, 1H), 2.02 (m, 1H), 1.94 (ddd, *J* = 14.4, 2.0 Hz, 1H), 1.71 (dt, *J* = 14.7, 9.5 Hz, 1H), 1.67–1.61 (m, 2H), 1.59 (dt, *J* = 14.4, 2.7 Hz, 1H), 1.54–1.44 (m, 3H), 1.41–1.36 (m, 2H), 1.32–1.20 (m, 10H), 0.88 (t, *J* = 6.8 Hz, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  169.3, 72.7, 72.5, 72.1, 67.0, 66.0, 43.2, 42.8, 37.7, 37.2, 36.4, 31.8, 29.6, 29.4, 29.3, 25.4, 22.6, 14.1; MS (ESI): m/z (%) 329 (30) [M+H]<sup>2</sup>, 351 (100) [M+Na]<sup>+</sup>; HRMS (ESI): calcd for C<sub>18</sub>H<sub>32</sub>O<sub>5</sub> Na [M+Na]<sup>+</sup> 351.2154, found 351.2147.